



Quantum Physics (量子物理) 習題

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CH 05 : Schroedinger's theory of quantum mechanics

5-01、If the wave function $\Psi_1(x,t)$, $\Psi_2(x,t)$, and $\Psi_3(x,t)$ are three solutions to the Schroedinger equation for a particular potential $V(x,t)$, show that the arbitrary linear combination $\Psi(x,t) = c_1\Psi_1(x,t) + c_2\Psi_2(x,t) + c_3\Psi_3(x,t)$ is also a solution to that equation.

<解> :

5-02、At a certain instant of time, the dependence of a wave function on position is as shown in Figure 5-20. (a) If a measurement that could locate the associated particle in an element dx of the x axis were made at that instant, where would it most likely be found? (b) Where would it least likely be found? (c) Are the chances better that it would be found at any positive value of x , or are they better that it would be found at any negative value of x ? (d) Make a rough sketch of the potential $V(x)$ which gives rise to the wave function. (e) To which allowed energy does the wave function correspond?

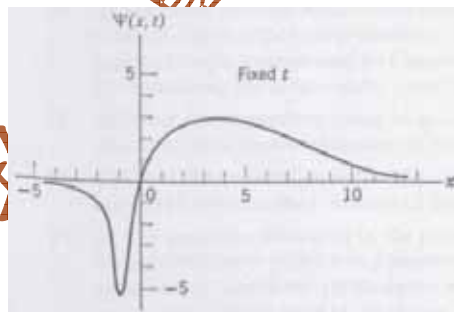


Figure 5-20 The space dependence of a wave function considered in Problem 2, evaluated at a certain instant of time.

<解> :

5-03、(a) Determine the frequency ν of the time-dependent part of the wave function quoted in Example 5-3, for the lowest energy state of a simple harmonic oscillator. (b) Use this value of ν , and the de Broglie-Einstein relation $E = h\nu$, to evaluate the total energy E of the oscillator. (c) Use this value of E to show that the limits of the classical motion of the oscillator, found in Example 5-6, can be





written as $x = \pm \frac{\hbar^{1/2}}{(Cm)^{1/4}}$.

<解> : (a) The time-dependent part of the wavefunction is $e^{-\frac{1}{2}it\sqrt{\frac{C}{m}}} = e^{-\frac{iEt}{\hbar}} = e^{-i2\pi\nu t}$

$$\text{Therefore, } \frac{1}{2}\sqrt{\frac{C}{m}} = 2\pi\nu \Rightarrow \nu = \frac{1}{4\pi}\sqrt{\frac{C}{m}}$$

$$(b) \text{ Since } E = h\nu = 2\pi\hbar\nu, \quad E = \frac{1}{2}\hbar\sqrt{\frac{C}{m}}$$

$$(c) \text{ The limiting } x \text{ can be found from } \frac{1}{2}Cx^2 = E$$

$$x = \pm\left(\frac{2E}{C}\right)^{1/2} = \pm\hbar^{1/2}(Cm)^{-1/2} \dots\dots\#\#$$

5-04、By evaluating the classical normalization integral in Example 5-6, determine the value of the constant B^2 which satisfies the requirement that the total probability of finding the particle in the classical oscillator somewhere between its limits of motion must equal one.

<解> : According to Example 5-6, the normalizing integral is

$$1 = 2B^2 \sqrt{\frac{m}{C}} \int_0^{\sqrt{\frac{2E}{C}}} \frac{dx}{\sqrt{\frac{2E}{C} - x^2}} = 2B^2 \sqrt{\frac{m}{C}} \sin^{-1} \frac{x}{\sqrt{\frac{2E}{C}}} \Bigg|_0^{\sqrt{\frac{2E}{C}}}$$

$$1 = B^2 \pi \sqrt{\frac{m}{C}} \Rightarrow B^2 = \left(\frac{C}{m\pi^2}\right)^{1/2} \dots\dots\#\#$$

5-05 Use the results of Example 5-5, 5-6, and 5-7 to evaluate the probability of finding a particle, in the lowest energy state of a quantum mechanical simple harmonic oscillator, within the limits of the classical motion. (Hint : (i) The classical limits of motion are expressed in a convenient form in the statement of Problem 3c. (ii) The definite integral that will be obtained can be expressed as a normal probability integral, or an error function . It can then be evaluated immediately by consulting mathematical handbooks which tabulate these quantities. Or, the integral can easily be evaluated by expanding the exponential as an infinite series before integrating, and then integrating the first few terms in the series. Alternatively, the definite integral can be evaluated by plotting the integrand on





graph paper, and counting squares to find the area enclosed between the integrand, the axis, and the limits.)

<解> : Problem 5-3(c) Provides the limits on x ; the wavefunction is

$$\Psi = \frac{(Cm)^{1/8}}{(\pi\hbar)^{1/4}} e^{-\frac{\sqrt{Cm}}{2\hbar}x^2} e^{-i\omega t}$$

Hence, the desired probability is given by

$$Prob. = 2 \frac{(Cm)^{1/4}}{(\pi\hbar)^{1/2}} \int_0^{\frac{\hbar^{1/2}}{(Cm)^{-1/4}}} e^{-\frac{\sqrt{Cm}}{\hbar}x^2} dx$$

$$\text{If } u = \frac{(4Cm)^{1/4}}{\hbar^{1/2}} x$$

$$Prob. = 2 \int_0^{\sqrt{2}} \frac{1}{\sqrt{2\pi}} e^{-\frac{u^2}{2}} du = 2(0.42) = 0.84 \dots\dots\#\#$$

5-06、At sufficiently low temperature, an atom of a vibrating diatomic molecule is a simple harmonic oscillator in its lowest energy state because it is bound to the other atom by a linear restoring force. (The restoring force is linear, at least approximately, because the molecular vibrations are very small.) The force constant C for a typical molecule has a value of about $C \sim 10^3 \text{ nt/m}$. The mass of the atom is about $m \sim 10^{-26} \text{ kg}$. (a) Use these numbers to evaluate the limits of the classical motion from the formula quoted in Problem 3c. (b) Compare the distance between these limits to the dimensions of a typical diatomic molecule, and comment on what this comparison implies concerning the behavior of such a molecule at very low temperatures.

<解>

5-07、(a) Use the particle in a box wave function verified in Example 5-9, with the value of A determined in Example 5-10, to calculate the probability that the particle associated with the wave function would be found in a measurement within a distance of $\frac{a}{3}$ from the right-hand end of the box of length a . The particle is in its lowest energy state. (b) Compare with the probability that would be predicted





classically from a very simple calculation related to the one in Example 5-6.

<解> : (a) Since $\Psi = \left(\frac{2}{a}\right)^{1/2} \cos \frac{\pi x}{a} e^{-\frac{iEt}{\hbar}}$

$$Prob. = \frac{2}{a} \int_{\frac{a}{6}}^{\frac{a}{2}} \cos^2 \left(\frac{\pi x}{a}\right) dx = \frac{2}{\pi} \int_{\frac{\pi}{6}}^{\frac{\pi}{2}} \cos^2 u du = \frac{1}{3} - \frac{\sqrt{3}}{4\pi} = 0.1955$$

Independent of E.

(b) Classically $Prob. = \frac{a/3}{a} = \frac{1}{3} = 0.3333 \dots \#\#$

5-08、Use the results Example 5-9 to estimate the total energy of a neutron of mass about $10^{-27} kg$ which is assumed to move freely through a nucleus of linear dimensions of about $10^{-14} m$, but which is strictly confined to the nucleus. Express the estimate in MeV. It will be close to the actual energy of a neutron in the lowest energy state of a typical nucleus.

<解> :

5-09、(a) Following the procedure of Example 5-9, verify that wave function

$$\Psi(x,t) = \begin{cases} A \sin \frac{2\pi x}{a} e^{-\frac{iEt}{\hbar}} & -\frac{a}{2} < x < +\frac{a}{2} \\ 0 & x < -\frac{a}{2} \text{ or } x > +\frac{a}{2} \end{cases}$$

is a solution to the schrodinger equation in the region $-\frac{a}{2} < x < +\frac{a}{2}$ for a particle which moves freely through the region but which is strictly confined to it.

(b) Also determine the value of the total energy E of the particle in this first excited state of the system, and compare with the total energy of the ground state found in Example 5-9. (c) Plot the space dependence of this wave function. Compare with the ground state wave function of Figure 5-7, and give a qualitative argument relating the difference in the two wave functions to the difference in the total energies of the two states.

<解> : (a) (b) Let $V = 0$ in the region in which the particle is confined, so that





Schrodinger's equation becomes $-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} = i\hbar \frac{\partial \Psi}{\partial t}$, with

$$\Psi = A \sin \frac{2\pi x}{a} e^{\frac{-iEt}{\hbar}}.$$

Putting these into Schrodinger's equation gives

$$\left(-\frac{\hbar^2}{2m}\right)\left(-\frac{4\pi^2}{a^2}\Psi\right) = i\hbar\left(-\frac{iE}{\hbar}\Psi\right) = E\Psi; \quad E = E_1 = \frac{2\pi^2\hbar^2}{ma^2}.$$

In the ground state, $E = E_0 = \frac{\pi^2\hbar^2}{2ma^2}$, so that $E = 4E_0$.

(c) The space parts of the wave functions are

$$\psi_0 = A \cos \frac{\pi x}{a}$$

$$\psi_1 = A \sin \frac{2\pi x}{a}$$

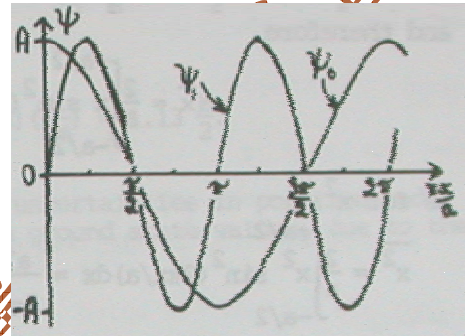
ψ_1 oscillates more rapidly, since with

$$E_1 > E_0,$$

$$\psi_1, \psi_0 \leq A,$$

$$\left|-\frac{d^2\psi_1}{dx^2}\right| = \frac{2m}{\hbar^2}|E_1\psi_1| > \left|-\frac{d^2\psi_0}{dx^2}\right| = \frac{2m}{\hbar^2}|E_0\psi_0|$$

for most x ##



5-10、(a) Normalize the wave function of Problem 9, by adjusting the value of the multiplicative constant A so that the total probability of finding the associated particle somewhere in the region of length a equals one. (b) Compare with the value of A obtained in Example 5-10 by normalizing the ground state wave function. Discuss the comparison.

<解> (a) To normalize the wavefunction, evaluate $1 = \int_{-\frac{a}{2}}^{\frac{a}{2}} \Psi^* \Psi dx$ ($\Psi = 0$ outside this region).

With $\Psi = A \sin \frac{2\pi x}{a} e^{\frac{-iEt}{\hbar}}$, this become

$$1 = 2A^2 \int_0^{\frac{a}{2}} \sin^2 \frac{2\pi x}{a} dx = \frac{a}{\pi} \int_0^{\pi} \sin^2 u du = \frac{a}{\pi} A^2 \frac{\pi}{2}$$





$$A = \sqrt{\frac{2}{a}} \dots\dots##$$

(b) This equals the value of A for the ground state wavefunction and, in fact, the normalization constant of all the excited states equals this also. Since all of the space wave functions are simple sines or cosines, this equality is understandable.....##

5-11、Calculate the expectation value of x , and the expectation value of x^2 for the particle associated with the wave function of Problem 10.

<解> : The wavefunction is $\psi = \sqrt{\frac{2}{a}} \sin \frac{2\pi x}{a} e^{-\frac{iEt}{\hbar}}$

$$\text{And therefore } \bar{x} = \frac{2}{a} \int_{-\frac{a}{2}}^{+\frac{a}{2}} x \sin^2 \frac{2\pi x}{a} dx = 0 \dots\dots##$$

As for x^2 :

$$\overline{x^2} = \frac{2}{a} \int_{-\frac{a}{2}}^{+\frac{a}{2}} x^2 \sin^2 \frac{2\pi x}{a} dx = \frac{a^2}{2\pi^3} \int_0^\pi u^2 \sin^2 u du = \frac{1}{4} \left(\frac{1}{3} - \frac{1}{2\pi^2} \right) a^2 = 0.07067 a^2 \dots\dots##$$

5-12、Calculate the expectation value of p , and the expectation value of p^2 , for the particle associated with the wave function of Problem 10.

<解> : The linear momentum operator is $-i\hbar \frac{\partial}{\partial x}$ and therefore

$$\bar{p} = \frac{2}{a} \int_{-\frac{a}{2}}^{+\frac{a}{2}} \sin \frac{2\pi x}{a} \left[-i\hbar \frac{\partial}{\partial x} \left(\sin \frac{2\pi x}{a} \right) \right] dx = -\frac{4i\hbar}{a} \int_0^\pi \sin u \cos u du = 0 \dots\dots##$$

Similarly,

$$\overline{p^2} = \frac{2}{a} \int_{-\frac{a}{2}}^{+\frac{a}{2}} \sin \frac{2\pi x}{a} \left[i^2 \hbar^2 \frac{\partial^2}{\partial x^2} \left(\sin \frac{2\pi x}{a} \right) \right] dx = -8\pi^2 i^2 \left(\frac{\hbar}{a} \right)^2 \int_0^\pi \sin^2 u du = 4\pi^2 \left(\frac{\hbar}{a} \right)^2 = \left(\frac{h}{a} \right)^2$$

.....##

5-13、(a) Use quantities calculated in the preceding two problems to calculate the





product of the uncertainties in position and momentum of the particle in the first excited state of the system being considered. (b) Compare with the uncertainty product when the particle is in the lowest energy state of the system, obtained in Example 5-10, Explain why the uncertainty products differ.

<解> : Let $\Delta x = \sqrt{x^2}$; $\Delta p = \sqrt{p^2}$.

(a) Problem 5-11 and 5-12 yield $\Delta x = na$, $n^2 = \frac{1}{4}(\frac{1}{3} - \frac{1}{2\pi^2})$; $\Delta p = \frac{h}{a}$.

Hence, $\Delta x \Delta p = (na)(\frac{h}{a}) = 4\pi n(\frac{\hbar}{2}) = (\frac{4}{3}\pi^2 - 2)^{1/2} \frac{\hbar}{2} = 3.34 \frac{\hbar}{2}$.

(b) In the ground state, $\Delta x \Delta p = (0.18a)(\frac{h}{2a}) = 1.13 \frac{\hbar}{2}$.

In the first excited state the uncertainties in position and momentum both increase over the ground state values, due to the higher energy of the particle.

5-14、(a) Calculate the expectation values of the kinetic energy and potential energy for a particle in the lowest energy state of a simple harmonic oscillator, using the wave function of Example 5-7. (b) Compare with the time-averaged kinetic and potential energies for a classical simple harmonic oscillator of the same total energy.

<解> : The normalized wavefunction is $\Psi = \frac{(Cm)^{1/8}}{(\pi\hbar)^{1/4}} e^{-\frac{\sqrt{Cm}x^2}{2\hbar}} e^{-\frac{iEt}{\hbar}}$ with $E = \frac{1}{2} \frac{\hbar}{(\frac{C}{m})}$

(a) Since the kinetic energy is $\frac{p^2}{2m}$ the corresponding operator is

$$T = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$$

$$\text{Therefore, } \bar{T} = \frac{(Cm)^{1/4}}{(\pi\hbar)^{1/2}} \left(-\frac{\hbar^2}{2m}\right) \int_{-\infty}^{+\infty} -\frac{\sqrt{Cm}}{2\hbar} x^2 \frac{\partial^2}{\partial x^2} e^{-\frac{\sqrt{Cm}x^2}{2\hbar}} dx$$

$$\bar{T} = \hbar \left(\frac{C}{\pi m}\right)^{1/2} \int_0^{\infty} (1-u^2) e^{-u^2} du = \frac{\hbar}{4} \left(\frac{C}{m}\right)^{1/2} = \frac{1}{2} E$$

Similarly for the potential energy $U = \frac{1}{2} Cx^2$





$$\bar{U} = \frac{(Cm)^{1/4}}{(\pi\hbar)^{1/2}} \frac{C}{2} \int_{-\infty}^{+\infty} x^2 e^{-\frac{\sqrt{Cm}x^2}{\hbar}} dx = \hbar \left(\frac{C}{\pi m}\right)^{1/2} \int_0^{\infty} u^2 e^{-u^2} du,$$

$$\bar{U} = \hbar \left(\frac{C}{\pi m}\right)^{1/2} \frac{\pi^{1/2}}{4} = \bar{T} = \frac{1}{2} E$$

(b) This same relation, $\bar{U} = \bar{T} = \frac{1}{2} E$, is obeyed by the classical oscillator also.....##

5-15、In calculating the expectation value of the product of position times momentum, an ambiguity arises because it is not approach which of the two expressions

$$\overline{xp} = \int_{-\infty}^{\infty} \Psi^* x \left(-i\hbar \frac{\partial}{\partial x}\right) \Psi dx$$

$$\overline{px} = \int_{-\infty}^{\infty} \Psi^* \left(-i\hbar \frac{\partial}{\partial x}\right) x \Psi dx$$

should be used. (In the first expression $\frac{\partial}{\partial x}$ operates on Ψ ; in the second it operates on $x\Psi$.) (a) Show that neither is acceptable because both violate the obvious requirement that \overline{xp} should be real since it is measurable. (b) Then show that the expression

$$\overline{xp} = \int_{-\infty}^{\infty} \Psi^* \left[\frac{x(-i\hbar \frac{\partial}{\partial x}) + (-i\hbar \frac{\partial}{\partial x})x}{2} \right] \Psi dx$$

is acceptable because it does satisfy this requirement. (Hint : (i) A quantity is real if it equals its own complex conjugate. (ii) Try integrating by part. (iii) In any realistic case the wave function will always vanish at $x = \pm\infty$.)

<解> :

5-16、Show by direct substitution into the Schroedinger equation that the wave function

$\Psi(x,t) = \psi(x)e^{-\frac{iEt}{\hbar}}$ satisfies that equation if the eigenfunction $\psi(x)$ satisfies the time-independent Schroedinger equation for a potential $V(x)$.





<解> :

5-17、(a) Write the classical wave equation for a string of density per unit length which varies with x . (b) Then separate it into two ordinary differential equations, and show that the equation in x is very analogous to the time-independent Schroedinger equation.

<解> :

5-18、By using an extension of the procedure leading to (5-31), obtain the Schroedinger equation for a particle of mass m moving in three dimensions (described by rectangular coordinates x,y,z)

<解> :

5-19、(a) Separate the Schroedinger equation of Problem 18, for a time-independent potential, into a time-independent Schroedinger equation and an equation for the time dependence of the wave function. (b) Compare to the corresponding one-dimensional equations, (5-37) and (5-38), and explain the similarities and the differences.

<解> :

5-20、(a) Separate the time-independent Schroedinger equation of Problem 19 into three time-independent Schroedinger equations, one in each of the coordinates. (b) Compare them with (5-37). (c) Explain clearly what must be assumed about the form of the potential energy in order to make the separation possible, and what the physical significance of this assumption is. (d) Give an example of a system that would have such a potential.

<解> :





5-21、Starting with the relativistic expression for the energy, formulate a Schrodinger equation for photons, and solve it by separation of variables, assuming $V = 0$.

<解> : With $V = 0$, the energy of the photon is $E = pc$.

Replacing the energy E and momentum p by their operators gives

$$i\hbar \frac{\partial \Psi}{\partial t} = -i\hbar c \frac{\partial \Psi}{\partial x}.$$

Now set $\Psi(x,t) = \psi(x)T(t)$ and divide the equation by ψ to get

$$i\hbar \frac{1}{T} \frac{dT}{dt} = -i\hbar c \frac{1}{\psi} \frac{d\psi}{dx} = K$$

Where K is independent of x and t . Write $K = k\hbar c$ and the two equations

$$\text{directly above become } \frac{dT}{dt} = -ikcT \Rightarrow T \propto e^{-ikct}$$

$$\frac{d\psi}{dx} = ik\psi \Rightarrow \psi = e^{ikx}$$

Hence, for the photon, $\Psi \propto e^{ik(x-ct)}$ ##

5-22、Consider a particle moving under the influence of the potential $V(x) = C|x|$,

where C is a constant, which is illustrated in Figure 5-21. (a) Use qualitative arguments, very similar to those of Example 5-12, to make a sketch of the first eigenfunction and of the tenth eigenfunction for system. (b) Sketch both of the corresponding probability density function. (c) Then use the classical mechanics to calculate, in the manner of Example 5-6, the probability density function predicted by that theory. (d) Plot the classical probability density functions with the quantum mechanical probability density functions, and discuss briefly their comparison.

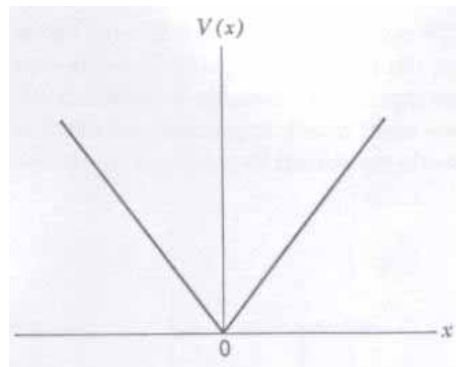


Figure 5-21 A potential function considered in Problem 22.



<解> :

- 5-23、Consider a particle moving in the potential $V(x)$ plotted in figure 5-22. For the following ranges of the total energy E , state whether there are any allowed values of E and if so, whether they are discretely separated or continuously distributed.
 (a) $E < V_0$, (b) $V_0 < E < V_1$, (c) $V_1 < E < V_2$, (d) $V_2 < E < V_3$, (e) $V_3 < E$.

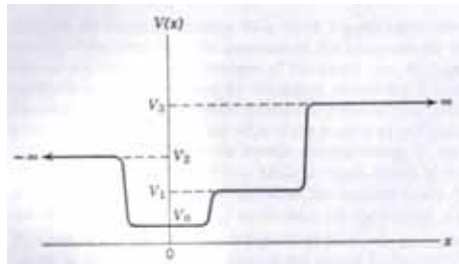


Figure 5-22 A potential function considered in Problem 23.

<解> :

- 5-24、Consider a particle moving in the potential $V(x)$ illustrated in Figure 5-23, that has a rectangular region of depth V_0 , and width a , in which the particle can be bound. These parameters are related to the mass m of the particle in such a way that the lowest allowed energy E_1 is found at an energy about $\frac{V_0}{4}$ above the “bottom.” Use qualitative arguments to sketch the approximant shape of the corresponding eigenfunction $\psi_1(x)$.

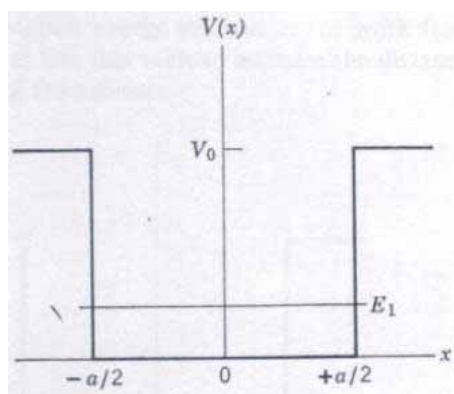


Figure 5-23 A potential function considered in problem24.

<解> :



5-25、 Suppose the bottom of the potential function of Problem 24 is changed by adding a bump in the center of height about $\frac{V_0}{10}$ and width $\frac{a}{4}$. That is, suppose the potential now looks like the illustration of Figure 5-24. Consider qualitatively what will happen to the curvature of the eigenfunction in the region of the bump, and how this will, in turn, affect the problem of obtaining an acceptable behavior of the eigenfunction in the region outside the binding region. From these considerations predict, qualitatively, what the bump will do to the value of the lowest allowed energy E_1 .

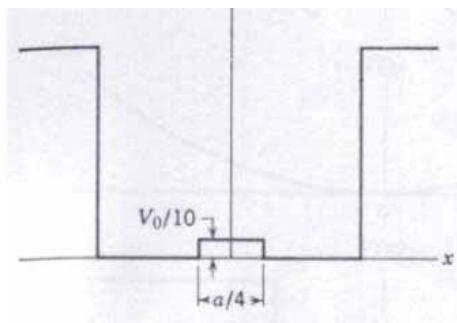


Figure 5-24 A rectangular bump added to the bottom of the potential of Figure 5-23; for Problem 25.

<解> : E_1 will increase

5-26、 Because the bump in Problem 25 is small, a good approximation to the lowest allowed energy of the particle in the presence of the bump can be obtained by taking it as the sum of the energy in the absence of the bump plus the expectation value of the extra potential energy represented by the bump, taking the Ψ corresponding to no bump to calculate the expectation value. Using this point of view, predict whether a bump of the same “size”, but located at the edge of the bottom as in Figure 5-25, would have a large, smaller, or equal effect on the lowest allowed energy of the particle, compared to the effect of a centered bump. (Hint : Make a rough sketch of the product of $\Psi^*\Psi$ and the potential energy function that describes the centered bump. Then consider qualitatively the effect of moving the bump to the edge on the integral of this product.)

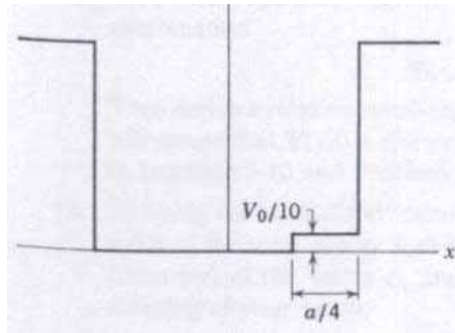


Figure 5-25 The same rectangular bump as in Figure 5-24, but moved to the edge of the potential ; for Problem 26.

<解> : smaller

5-27 、 By substitution into the time-independent Schroedinger equation for the potential illustrated in Figure 5-23, show that in the region to the right of the binding region the eigenfunction has the mathematical form

$$\psi(x) = Ae^{-\frac{\sqrt{2m(V_0-E)}x}{\hbar}}, x > +\frac{a}{2}.$$

<解> : Schroedinger's equation is

$$\frac{d^2\psi}{dx^2} + \frac{2m}{\hbar^2}(E-V)\psi = 0$$

In the region in question, $V = V_0 = \text{constant}$, $E < V_0$, so that

$$q^2 = \frac{2m}{\hbar^2}(V_0 - E) > 0$$

Hence, $\psi = Ae^{-qx} + Be^{qx}$, is the general solution. However, $\psi(x = \infty) = 0$, requiring $B = 0$

$\psi = Ae^{-qx}$ as the wavefunction.....##

5-28 、 Using the probability density corresponding to the eigenfunction of Problem 27, write an expression to estimate the distance D outside the binding region of the potential within which there would be an appreciable probability of finding the particle. (Hint : Take D to extend to the point at which $\Psi^*\Psi$ is smaller than its value at the edge of the binding region by a factor of e^{-1} . This e^{-1} criterion is similar to one often used in the study of electrical circuits.)

<解> : Since ψ is real, the probability density P is $P = \psi^*\psi = \psi^2 = A^2e^{-2qx}$



Recalling that x is measured from the center of the binding region, the suggested

criterion for D gives $A^2 e^{-2q(\frac{1}{2}a+D)} = e^{-1} A^2 e^{-2q(\frac{1}{2}a)}$

$$e^{qa-2qD} = e^{-qa-1}$$

$$D = \frac{1}{2q} = \frac{\hbar}{2[2m(V_0 - E)]^{1/2}} \dots\dots##$$

- 5-29、The potential illustrated in Figure 5-23 gives a good description of the forces acting on an electron moving through a block of metal. The energy difference $V_0 - E$, for the highest energy electron, is the work function for the metal. Typically, $V_0 - E \approx 5\text{eV}$. (a) Use this value to estimate the distance D of Problem 28. (b) Comment on the results of the estimate.

<解> : From Problem 28

$$D = \frac{\hbar}{2[2m(V_0 - E)]^{1/2}} = \frac{\hbar}{2[2(9.1 \times 10^{-31})(5)(1.6 \times 10^{-19})]^{1/2}} = 0.4 \text{ \AA}$$

- 5-30、Consider the eigenfunction illustrated in the top part of Figure 5-26. (a) Which of the three potentials illustrated in the bottom part of the figure could lead to such an eigenfunction? Give qualitative arguments to justify your answer. (b) The eigenfunction shown is not the one corresponding to the lowest allowed energy for the potential. Sketch the form of the eigenfunction which does correspond to the lowest allowed energy E_1 . (c) Indicate on another sketch the range of energies where you would expect discretely separated allowed energy states, and the range of energies where you would expect the allowed energies to be continuously distributed. (d) Sketch the form of the eigenfunction which corresponds to the second allowed energy E_2 . (e) To which energy level does the eigenfunction presented in Figure 5-26 correspond?



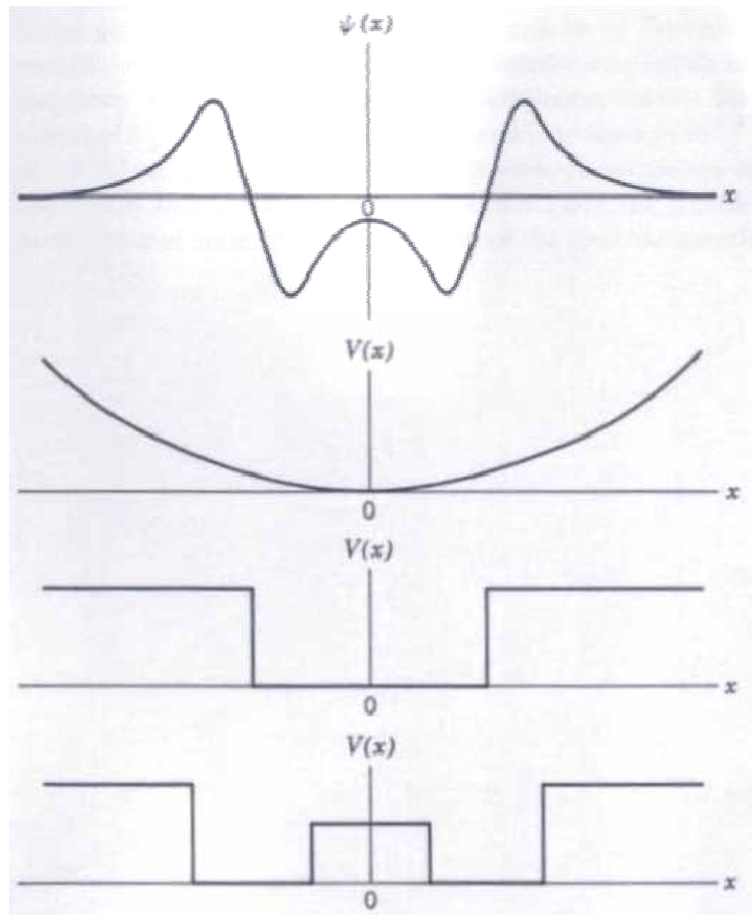


Figure 5-26 An eigenfunction (top curve) and three possible forms (bottom curves) of the potential energy function considered in Problem 30.

<解> :

5-31、Estimate the lowest energy level for a one-dimensional infinite square well of width a containing a cosine bump. That is the potential V is

$$V = V_0 \cos \frac{\pi x}{a} \quad -\frac{a}{2} < x < +\frac{a}{2}$$

$$V = \text{infinity} \quad x < -\frac{a}{2} \quad \text{or} \quad x > +\frac{a}{2}$$

where $V_0 \ll \frac{\pi^2 \hbar^2}{2ma^2} \dots \dots \# \#$

<解> :





5-32、Using the first two normalized wave function $\Psi_1(x,t)$ and $\Psi_2(x,t)$ for a particle moving freely in a region of length a , but strictly confined to that region, construct the linear combination $\Psi(x,t) = c_1\Psi_1(x,t) + c_2\Psi_2(x,t)$. Then derive a relation involving the adjustable constants c_1 and c_2 which, when satisfied, will ensure that $\Psi(x,t)$ is also normalized. The normalized $\Psi_1(x,t)$ and $\Psi_2(x,t)$ are obtained in Example 5-10 and Problem 10.

<解> :

5-33、(a) Using the normalized “mixed” wave function of Problem 32, calculate the expectation value of the total energy E of the particle in terms of the energies E_1 and E_2 of the two states and the values c_1 and c_2 of the mixing parameters. (b) Interpret carefully the meaning of your result.

<解> : (a) The total energy is $E = \frac{p^2}{2m} + V$. But $V \neq 0$ in the region of motion, so that

$$E = \frac{p^2}{2m} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$$

$$\text{Hence, } \bar{E} = -\frac{\hbar^2}{2m} \int_{-\frac{a}{2}}^{+\frac{a}{2}} (c_1^*\Psi_1^* + c_2^*\Psi_2^*) \frac{\partial^2}{\partial x^2} (c_1\Psi_1 + c_2\Psi_2) dx.$$

$$\text{But } \frac{\partial^2\Psi_1}{\partial x^2} = -\left(\frac{\pi}{a}\right)^2\Psi_1; \quad \frac{\partial^2\Psi_2}{\partial x^2} = -\left(\frac{2\pi}{a}\right)^2\Psi_2.$$

Also, by Problem 5-32, $\int \Psi_1^*\Psi_2 dx = \int \Psi_2^*\Psi_1 dx = 0$ and therefore

$$\bar{E} = \frac{\hbar^2}{2m} \left\{ \left(\frac{\pi}{a}\right)^2 c_1 c_1^* \int \Psi_1^*\Psi_1 dx + \left(\frac{2\pi}{a}\right)^2 c_2 c_2^* \int \Psi_2^*\Psi_2 dx \right\},$$

$$\bar{E} = c_1 c_1^* \frac{\pi^2 \hbar^2}{2ma^2} + c_2 c_2^* \frac{2\hbar^2 \pi^2}{ma^2}$$

$$\bar{E} = c_1 c_1^* E_1 + c_2 c_2^* E_2$$

(b) Since $c_1 c_1^* + c_2 c_2^* = 1$, $\bar{E} = (1 - c_2 c_2^*) E_1 + c_2 c_2^* E_2 = E_1 + c_2 c_2^* (E_2 - E_1)$

With $0 \leq c_2 c_2^* \leq 1$, this means that $E_1 \leq \bar{E} \leq E_2$.

Hence, if the particle can be found either in level 1 or 2, making transitions between them, its average energy, as would be expected, lies between the





energies of the two levels.....##

5-34 · If the particle described by the wave function of Problem 32 is a proton moving in a nucleus, it will give rise to a charge distribution which oscillates in time at the same frequency as the oscillations of its probability density. (a) Evaluate this frequency for values of E_1 and E_2 corresponding to a proton mass of 10^{-27} kg and a nuclear dimension of 10^{-14} m. (b) Also evaluate the frequency and energy of the photon that would be emitted by oscillating charge distribution as the proton drops from the excited state to the ground state. (c) In what region of the electromagnetic spectrum is such a proton?

<解> : (a) The probability density $\Psi^*\Psi$ has a time dependence of $e^{-i(E_2-E_1)t/\hbar}$, and

therefore the frequency is $\nu = \frac{E_2 - E_1}{h}$

$$\text{But, } E_1 = \frac{\pi^2 \hbar^2}{2ma^2} = \frac{h^2}{8ma^2} = \frac{(6.626 \times 10^{-34})^2}{8(1.67 \times 10^{-27})(10^{-14})^2(1.602 \times 10^{-13})}$$

$$E_1 = 2.051 \text{ MeV}; \quad E_2 = 4E_1 = 8.204 \text{ MeV}.$$

$$\text{Hence, } \nu = \frac{8.204 - 2.051}{4.136 \times 10^{-15}} = 1.488 \times 10^{21} \text{ Hz}.$$

(b) The frequency of the photon is the same as in (a). The photon's energy is

$$h\nu = 8.204 - 2.051 = 6.153 \text{ MeV}$$

(c) Photons with this energy lie in the gamma-ray region of the spectrum.....##

